



## Article

# Hydrogel Formulation of Usnic Acid and Antibacterial Activity Test Against *Propionibacterium acne*

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Received: 8 November 2018; Accepted: 17 December 2018; Published: 20 December 2018



**Abstract:** Usnic acid is known for its remarkable antimicrobial activity. The aim of this research was to formulate hydrogel of usnic acid and evaluate the antibacterial activity against *Propionibacterium acne*. Due to low solubility of usnic acid, solid dispersion was prepared using PVP K-30. In this study, intact usnic acid (UA) and usnic acid-solid dispersion (UA-SD) was formulated in hydrogel using several gelling agents: Aqupec HV-505, sodium alginate and HPMC K 100M. Concentration of each gelling agent was optimized for hydrogel base. All of hydrogel base showed homogenous gel, pH at range 5.37–6.33 and viscosity in range 259.07–10,759.00 cps. Hydrogel was prepared by dispersing 1% intact UA and 3% UA-SD in three different gelling agents. The hydrogel was evaluated for pH, viscosity, stability test for two months and microbiology test. The amount of usnic acid in hydrogel was determined by spectrophotometry UV-Vis. Hydrogel UA showed non-homogenous gel, while hydrogel usnic UA-SD was homogenous. The pH of all hydrogel was in range 5.5–6.4 and viscosity was 2,017.03–3,866.52 cps. All the hydrogel was stable and diameter inhibition of hydrogel was in a range 20–32 mm. The amount of usnic acid in hydrogel was in range 96.9–99.23%. In conclusion, hydrogel UA-SD is promising preparation in handling acne.

**Keywords:** usnic acid; hydrogel; *Propionibacterium acne*; solid dispersion

## 1. Introduction

Usnic acid, a secondary metabolite produced by *Usnea* sp, is already known for remarkable pharmacological activities including: anti-bacteria [1], antiviral [2], anti-proliferation [3], anticancer [4], antioxidant [5], antipyretic and analgesic [6] and anti-inflammation [7]. Usnic acid has two optical isomers, the (+) and (−) usnic acid, which affected its activity against microorganisms. Both of enantiomers are potent against Gram-positive bacteria, while the (−) usnic acid has typical activity as a natural herbicide [8]. Moreover, usnic acid also has a significant activity against anaerobic Gram positive bacteria including *Propionibacterium* species [9,10]. *Propionibacterium acne*, a Gram-positive and facultative anaerobic bacterium, is one of the normal flora on the skin that in certain conditions can cause inflammation by producing lipases which break down fatty acids free of skin lipids, known as acnes [11]. A study has confirmed that usnic acid from *Usnea barbata* has ability to inhibit the growth of *P. acnes* at concentration  $\geq 1 \mu\text{g/mL}$  [12].

There are numerous activities of usnic acid, yet its availability in the market for pharmaceutical products is still limited. In Indonesia, there is only one commercial product that has been

marketed—Cream Scabid<sup>®</sup> which contained 1% usnic acid. This finite number of product is likely due to low solubility of usnic acid in water (0.01 g/100 mL) [8]. Some efforts have been done to increase the solubility of usnic acid, such as complex formation using cyclodextrin [13], microencapsulation using polymer PLGA [14], milling process [15] and preparation in solid dispersion using HPMC [16]. Our recent study has shown enhancement of usnic acid by preparing it in solid dispersion using poly-vinyl-pirolidon (PVP) K-30 and its potency as an anti-oxidant was in accordance with the solubility result [17].

To the best of our knowledge, there was only one application of usnic acid as an anti-acne that has been investigated and patented by converting usnic acid into salt form or metal salt [18]. Meanwhile, one of the most favourable preparations used for handling acne is hydrogel. Hydrogel offers convenience in use and has more attractive appearance due to its transparency compared to other preparations. Moreover, hydrogel is easily washed out and give cool sensation during application on skin due to water content of the base. Therefore, gelling agents, such as chitosan, HPMC, carbomer, HPMC, polyvinyl alcohol (PVA) and sodium alginate, play important part in preparing hydrogel [12]. As a preliminary study, this research was focused on the pharmaceutical aspects of hydrogel contained usnic acid and the potency of usnic acid in order to confirm the activity against *P. acnes*.

Based on the above considerations, preparation of hydrogel containing intact usnic acid and usnic acid-solid dispersion was carried out in this research. Solid dispersion was prepared using PVP K-30 in order to increase the solubility of usnic acid [17]. Prior to preparation of usnic acid hydrogel, concentration of gelling agents used is optimized. Hydrogel of usnic acid was then evaluated by homogeneity, pH, stability and anti-bacterial test against *P. acnes*.

## 2. Materials and Methods

### 2.1. Materials

(+) Usnic acid (isolated from *Usnea* sp as explained in previous work [13]), PVP K-30 (Shin-Etsu Chemical, Tokyo, Japan), Aqupec HV-505 (Sumitomo Seika Chemicals Co., Ltd., Osaka, Japan), sodium alginate (PT. Kimia Farma, Jakarta, Indonesia), HPMC K 100 M (PT. Kimia Farma, Jakarta, Indonesia), glycerin (Bratachem, Jakarta, Indonesia), triethanolamine (Bratachem, Jakarta, Indonesia), nutrient agar (Merck, Darmstadt, Germany), *Propionibacterium acnes* ATCC 6919 (The Laboratory of Natural Resource of Sumatra, Padang, Indonesia), clindamycin phosphate gel (Medi-Klin, PT. Surya Dermato Medika, Surabaya, Indonesia), phosphate buffer (Merck, Darmstadt, Germany), chloroform (Merck, Darmstadt, Germany), ethanol (Bratachem, Jakarta, Indonesia) and distilled water.

### 2.2. Optimization of Gelling Agent for Hydrogel Base

Formulation of hydrogel base was prepared using three different gelling agents which were Aqupec 505 HV, Sodium alginate and HPMC K 100M at different concentration, as seen in Table 1. Hydrogel base was prepared by dispersing the gelling agent in distilled water until its swelling well.

### 2.3. Evaluation of Hydrogel Base

Each hydrogel base was evaluated using several parameters included: organoleptic test by observed its appearance visually; homogeneity test by dispersed about 1 g of each hydrogel sample was on an object glass and observed its homogeneity; pH measurement by diluted hydrogel into 1% concentration and pH was determined using pH meter digital (Hanna Instruments, Woonsocket, RI, USA); viscosity test by using a Brookfield viscometer (Ametek, Berwyn, PA, USA); and wash out test by applying 1 g each hydrogel to the hand then washed by water.

**Table 1.** Preparation of hydrogel base.

Materials	Formula 1			Formula 2			Formula 3		
	1a	1b	1c	2a	2b	2c	3a	3b	3c
Aqupec HV–505 (%)	0.15	0.175	0.20	—	—	—	—	—	—
Sodium alginate (%)	—	—	—	1.5	1.75	2	—	—	—
HPMC K 100 M (%)	—	—	—	—	—	—	1.5	1.75	2
Glycerin (%)	10	10	10	10	10	10	10	10	10
TEA (%)	0.4	0.4	0.4	—	—	—	—	—	—
Distilled water ad (%)	100	100	100	100	100	100	100	100	100

a, b, and c are the variation of gelling agents concentration used in this study.

#### 2.4. Preparation Solid Dispersion

Solid dispersion of usnic acid PVP K-30 was prepared as described in previous work [15]. Usnic acid and PVP K-30 at ratio 1:2 (*w/w*) were dispersed in 100 mL distilled water and stirred until homogenous. The mixture was then frozen using liquid nitrogen prior to drying process and continued to primary and secondary drying in freeze dryer apparatus (Christ Alpha 1-2 LD Plus, Osterode, Germany). The dried powder was then kept in a desiccator.

#### 2.5. Preparation and Evaluation of Hydrogel Usnic Acid (UA) and Usnic Acid in Solid Dispersion (UA-SD)

The optimum concentration of each gelling agent was prepared for hydrogel containing intact usnic acid and usnic acid in solid dispersion. The amount of usnic acid dispersed in hydrogel was equivalent to 1% (*w/w*), so that the amount of usnic acid-solid dispersion was 3% (*w/w*). Each of optimal concentration of gelling agent was used in this hydrogel preparation. Furthermore, hydrogel preparations were evaluated for organoleptic test, homogeneity, pH and viscosity as described in evaluation of hydrogel base.

#### 2.6. Spreadability and Syneresis Test

Spreadability test was conducted by placing about 0.5 g of hydrogel on transparent glass equipped with on a graph paper. The hydrogel was covered with transparent plastic and given a certain load (1, 3, 5 and 7 g) for 15 s. The diameter was measured after being given a load. In addition, syneresis test was done by storing the hydrogel at 10 °C for 24, 48 and 72 h. Each hydrogel was placed on a cup to hold water released from the gel during storage. Syneresis value was calculated by measuring the weight loss during storage then compared to the initial weight.

#### 2.7. Usnic Acid Assay in Hydrogel

The amount of usnic acid in hydrogel was determined by a spectrophotometer UV-Vis (Shimadzu, Japan) at maximum absorption length in phosphate buffer. About 1 g of each hydrogel was dissolved with 100 mL phosphate buffer pH 7.4 and homogenized by sonication for 2 h. The sample was then filtered and measured the absorbance at 289 nm. The amount of usnic acid was then calculated based on linear regression.

#### 2.8. Cycling and Stability Test

Cycling test: The hydrogel was tested for stability against cooling condition. Each sample was stored at temperature of 0–4 °C for 24 h. The homogeneity of each sample was observed and pH was determined after the cycling test.

Stability test: Each sample of hydrogel was kept in at room temperature for 8 weeks. The stability of sample was observed every week including the physical appearance, homogeneity and pH test.

### 2.9. In Vitro Antibacterial Activity

One ml of *Propionibacterium acne* suspension stock was put into a sterilized petri dish and added nutrient agar medium. The mixture was homogenized. About 10 mg of each hydrogel sample was then put on the wells that has been prepared and incubated for 24 h at 37 °C. The ability of hydrogel samples to inhibit the growth of *Propionibacterium acne* was determined by measuring diameter inhibition. As positive and negative controls, a marketed gel containing clindamycin phosphate 1.2% (Medi-klin) and hydrogel base were tested using the same procedure.

## 3. Results

### 3.1. Optimization of Hydrogel Base

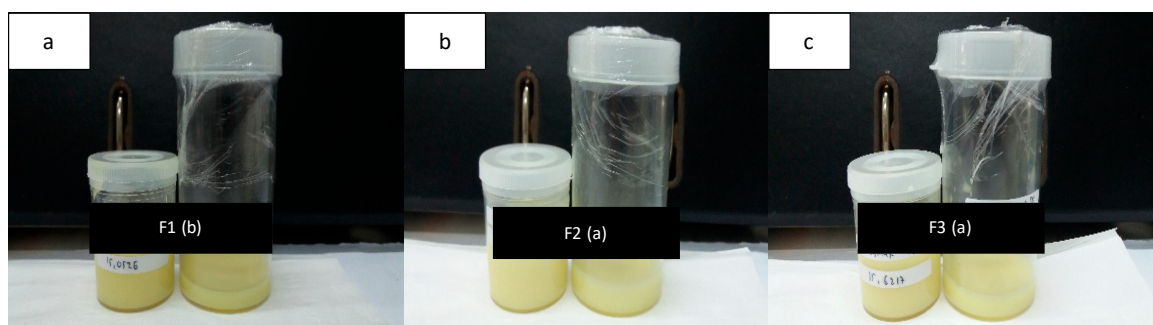
The result of hydrogel base optimization is shown in Table 2. All the hydrogel showed transparent and homogenous gel. Moreover, hydrogel had pH in a range 5.4–6.4 which is acceptable in preparing hydrogel. However, the viscosity of the F1a was the lowest which was relatively as liquid than gel. This result was likely due to insufficient amount of gelling agent used to prepare the hydrogel as described in product description [19], thus the concentration of Aqupec hydrogel influenced the viscosity. The wash out result denoted that the higher concentration of gelling agent, the more water need to wash the gel out of the skin. Thus, preparation of hydrogel containing intact usnic acid (UA) and usnic acid-solid dispersion (UA-SD) was carried out using F1b, F2a and F3a.

**Table 2.** Result of hydrogel base optimization.

No.	Hydrogel Base	Homogeneity	pH	Volume of Wash out	Viscosity
1	F1a	Homogenous	6.33	16 mL	259.07 cps
2	F1b	Homogenous	6.20	17 mL	2595.39 cps
3	F1c	Homogenous	6.53	18 mL	3703.75 cps
4	F2a	Homogenous	5.40	32 mL	3815.19 cps
5	F2b	Homogenous	5.37	37 mL	5115.17 cps
6	F2c	Homogenous	6.00	40 mL	14,203.22 cps
7	F3a	Homogenous	5.80	40 mL	5544.87 cps
8	F3b	Homogenous	5.40	50 mL	9811.03 cps
9	F3c	Homogenous	5.40	60 mL	15,811.03 cps

### 3.2. Hydrogel of Usnic Acid and Usnic Acid—Solid Dispersion

The hydrogel contained usnic acid and usnic acid-solid dispersion showed a yellowish gel which is shown in Figure 1. The result of homogeneity, pH, viscosity and usnic acid content in hydrogel usnic acid and usnic acid-solid dispersion is shown in Table 3. The hydrogel of intact usnic acid (UA) was not homogenous while usnic acid-solid dispersion (UA-SD) was homogenous. This result has been predicted since the intact usnic acid crystal could not be well dispersed in hydrogel base. Meanwhile, the addition of usnic acid in hydrogel influence the pH result. Usnic acid is a weak acid that has pKa 4.4 [20]. Moreover, PVP K-30 was used to enhance the solubility of usnic acid [17], which impacted in reducing the viscosity of hydrogel. This phenomenon was also observed in other study [21,22], in which PVP maintains adhesion and limits of the physiological range. The amount of usnic acid in hydrogel was almost close to each formula which indicated that all intact usnic acid and usnic acid-solid dispersion has the same amount in hydrogel preparation.



**Figure 1.** Hydrogels of usnic acid in different gelling agent (a) Aquapex, (b) sodium alginate, (c) HPMC.

**Table 3.** Result of hydrogel test.

Formula	Homogeneity	pH	Viscosity (cps)	Usnic Acid Assay (%)
F1b UA	No	5.63 ± 0.09	2328.67	97.56 ± 0.19
F1b UA-SD	Homogeneous	5.67 ± 0.05	2270.74	99.23 ± 0.29
F2a UA	No	5.70 ± 0.08	2038.95	96.94 ± 0.51
F2a UA-SD	Homogeneous	5.50 ± 0.08	2017.03	98.19 ± 0.10
F3a UA	No	5.67 ± 0.05	3612.63	96.90 ± 0.29
F3a UA-SD	Homogeneous	5.50 ± 0.14	3635.04	97.77 ± 0.29

Spreadability and syneresis test of hydrogel UA and UA-SD were also conducted to characterize the hydrogel as seen in Table 4. The spreadability value was corresponding to the viscosity result in previous evaluation. According to the result, the lower viscosity of hydrogel the greater ability to spread over. Meanwhile, syneresis is the ability of gel to be shrinkage due to change in temperature. The syneresis result of all hydrogels were similar.

**Table 4.** Spreadability and syneresis test result.

Formula	Spreadability (cm)					Syneresis (g)		
	Load 0 g	Load 1 g	Load 3 g	Load 5 g	Load 7 g	24 h	48 h	72 h
F1b UA	1.90	2.30	2.75	3.25	3.90	11.20	11.13	11.11
F1b UA-SD	2.85	3.45	3.95	4.45	5.10	11.18	11.15	11.14
F2a UA	1.85	2.35	2.70	2.85	3.00	11.05	11.37	11.31
F2a UA-SD	2.05	2.45	2.70	2.95	3.10	11.03	11.03	10.93
F3a UA	2.20	2.70	3.00	3.20	3.40	11.02	11.95	10.95
F3a UA-SD	2.40	2.85	3.15	3.55	3.65	11.04	10.94	11.01

The result of cycling test in Table 5 indicated the stability of hydrogel under cool temperature, while stability of hydrogel at room temperature for eight weeks can be seen in Table 6. The cycling test did show similar result of two parameters before and after the test. Similarly, the stability test was relatively the same after eight-week storage which suggested all hydrogel was stable.

**Table 5.** Cycling test result.

Formula	pH		Homogeneity	
	Before	After	Before	After
F1b UA	5.63 ± 0.09	5.67 ± 0.00	No	No
F1b UA-SD	5.67 ± 0.05	5.67 ± 0.03	Homogeneous	Homogeneous
F2a UA	5.70 ± 0.08	5.63 ± 0.05	No	No
F2a UA-SD	5.50 ± 0.08	5.47 ± 0.05	Homogeneous	Homogeneous
F3a UA	5.67 ± 0.05	5.63 ± 0.05	No	No
F3a UA-SD	5.50 ± 0.14	5.20 ± 0.02	Homogeneous	Homogeneous



Table 6. Stability test at room temperature.

Formula	pH Homogeneity			
	Week 1	Week 2	Week 4	Week 8
F1b	6.30 Homogeneous	6.20 Homogeneous	6.20 Homogeneous	6.20 Homogeneous
F1b UA	5.63 No	5.81 No	5.83 No	5.83 No
F1b UA-SD	5.67 Homogeneous	5.63 Homogeneous	5.63 Homogeneous	5.63 Homogeneous
F2a	6.60 Homogeneous	6.30 Homogeneous	6.00 Homogeneous	6.00 Homogeneous
F2a UA	5.50 No	5.50 No	5.50 No	5.47 No
F2a UA-SD	5.70 Homogeneous	5.70 Homogeneous	5.70 Homogeneous	5.70 Homogeneous
F3a	5.83 Homogeneous	5.73 Homogeneous	5.73 Homogeneous	5.73 Homogeneous
F3a UA	5.50 No	5.43 No	5.43 No	5.43 No
F3a UA-SD	5.57 Homogeneous	5.57 Homogeneous	5.57 Homogeneous	5.57 Homogeneous

### 3.3. In Vitro Antibacterial Assay

The result of previous evaluations was anticipated to the antibacterial activity of usnic acid in hydrogel preparation. The diameter zone inhibition of each sample was correlated to the activity of antibacterial assay against *P. acne*, which is shown in Table 7 and Figure 2.

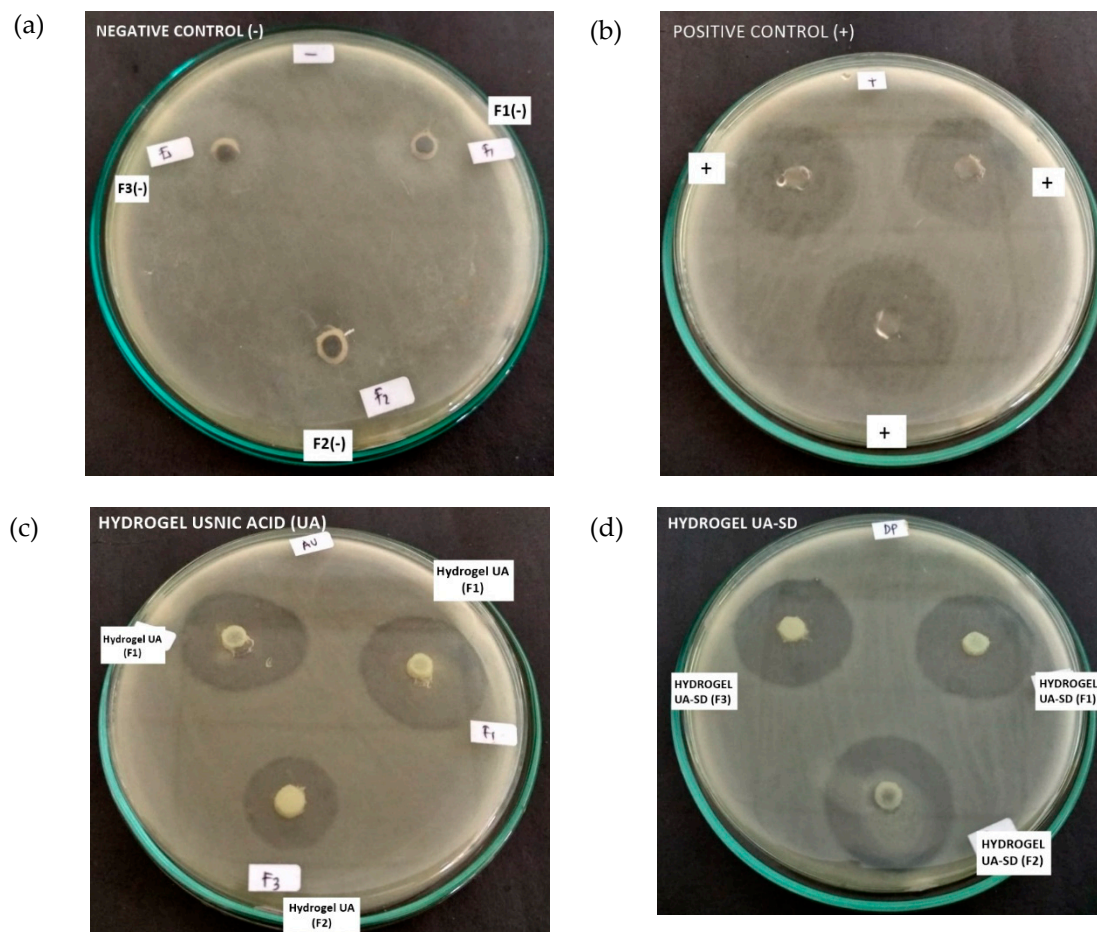


Figure 2. Result of antibacterial test against *Propionibacterium acne*.

**Table 7.** Result of anti-bacterial assay.

Formula	Diameter Inhibition (mm) $\pm$ SD
F1b	0 $\pm$ 0.00
F1b UA	30 $\pm$ 0.05
F1b UA-SD	32 $\pm$ 0.05
F2a	0 $\pm$ 0.00
F2a UA	29 $\pm$ 0.05
F2a UA-SD	30 $\pm$ 0.05
F3a	0 $\pm$ 0.00
F3a UA	20 $\pm$ 0.05
F3a UA-SD	26 $\pm$ 0.05
Positive control (+)	30 $\pm$ 0.05

All the hydrogel base or negative control showed no inhibition zone which indicated that no activity of antibacterial activity, see Figure 2a. Meanwhile, the positive control which used clindamycin gel shows about 30 mm diameter zone inhibition, see Figure 2b. The largest diameter inhibition was the hydrogel contained usnic acid-solid dispersion using Aqupec as the gelling agent. This result pointed that the hydrogel of usnic acid which prepared with different gelling agents had the same activity against *P. acnes* compared to the marketed and synthetic antibiotics. The potency of usnic acid as antibacterial is owing to the phenolic groups that known to have antibacterial activity [23]. In addition, the enhancement of antimicrobial activity in solid dispersion against bacteria was also showed in solid dispersion of usnic acid-polyacrilamide [24].

#### 4. Conclusions

Preparation of hydrogel is influenced by the form of usnic acid used, which solid dispersion of usnic acid with PVP K-30 provides better result in appearance compared to intact usnic acid. PVP K-30 has not only increased the solubility of usnic acid but also influenced the homogeneity of usnic acid in hydrogel. The hydrogels were relatively stable after eight-weeks storage. The antibacterial activity against *Propionibacterium acne* by hydrogel of usnic acid and usnic acid-solid dispersion was almost similar to marketed gel contained synthetic antibiotic, which is a promising result for further study.

**Author Contributions:** Design this research was led by A.B. The preparation and evaluation of hydrogel was done by Afifah. The isolation of usnic acid from *Usnea* sp was done by A.B. and F.I. The data were analysed by L.F. and Afifah. The manuscript was written by L.F., Afifah and F.I.

**Funding:** This research was funded by Andalas University, grant number No. 24/UN.16.17/PP.RGB/LPPM/2018.

**Acknowledgments:** We gratefully acknowledge Andalas University for financial support and Dr. Erizal Zaini for his contribution in preparing solid dispersion of usnic acid.

**Conflicts of Interest:** The authors declare no conflict of interest. The funders had no role in the design of the study; in the collection, analyses or interpretation of data; in the writing of the manuscript or in the decision to publish the results.

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